

How chemistry can help Brown Stock Washing operations

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Water is the connection **Kemira**

"Air is our enemy"



The Effect of Air Content for Washing Capacity

A high air content of pulp and wash water will have negative effect on washing capacity

Already an air content of a few percent in the pulp will significantly decrease the washing capacity of the washer

Relative capacity of a wash filter vs. air content in the pulp (drum washer)



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How Defoamer Works





Deaeration





Where Entrained Air and Foam are coming from?

Softwood	Hardwood
High foaming tendency	Lower foaming tendency
5 – 8 % extractives	2-4 % overall extractives. (Maple and aspen 6 -7 %)
Higher free acid content due to resin acids	More esterified fattyacids.

Mixing of air into stock or liquor!

- Filtration at washers
- Too low vat levels in washing filters
- Too low levels in filtrate tanks.
- Flashing because of too high temperature. (Boyle's law works also in practice)
- Etc...

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What are the Recommended Dosing Points?

- Before process point where air can escape from the liquor at atmospheric pressure.
- Dosing to liquor/filtrate is usually better, because consistency of the pulp can be quite high.
- Better to have more feed in dosage to the first washers than at later stages.
- Dose at least 60% to the first washing stages. Mostly 70 80 %.
- If there is a problem with shower displacement, dosing can be done to the shower water.
- Dosing point to O2 blow tube.



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How to choose a defoamer? Kemira's FEAT method





Principle of defoamer testing



- 1. Pumping started
- 2. Foam generated
- 3. Defoamer added





Performance of defoamers

Defoamer chemistry

- Type of defoamer; oil, silicone, waterbase...
- Formulation (raw materials composition)
- Colloidal characteristics (e.g. particle size)

Industrial application

- Pulp mills (black liquor, brownstock)
 - HW / SW liquors different soaps
 - Temperature and alkali
 - Dissolved solids
- Paper mills
 - Paper grade & furnish type; extractives, dissolved colloids, calcium soaps, dissolved carbonate & microbubbles...



Selection of defoamer:

- always process-specific
- · always to be optimized





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Development of FEAT method

FEAT provides responses of defoamers

- Informative from application point of view, i.e. FEAT provides enough data for comparison of products; "foam killing efficiency"
- The relationship between chemical properties of process liquors and FEAT response curves is incomplete → there is need to measure other properties of liquors
 - Dissolved solids
 - Conductivity
 - Dissolved ions; Na, K, Ca, SO4...
 - O2, pH, redox (mainly wastewater applications)
 - Refractive Index: total concentration of dissolved substances







Optimized defoamers:

- Chemistry
- **Formulation**
- Colloidal characteristics

From product testing to online control





Foaming \rightarrow

- 1. Decreased washing efficiency
- 2. Increased carry-over





From product testing to online control



Defoaming →

- 1. Measurement of entrained gas
- 2. Measurement of dissolved substances
- 3. Online control of defoamer dosage





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CONCLUSIONS

Brownstock operations are always impaired by

- Air, which is always present
- Overrun capacity

Chemistry can help alleviate these drawbacks

Creating a succesful chemistry solution is a function of:

- 1. Proper chemistry testing protocol and product proposal
- 2. Proper system survey with sophisticated control proposal
 - Here proper instruments come into play

